

# Applications of Predictive Stability in Early Development

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Science of Stability, 3<sup>rd</sup> Annual Conference, Dublin

4<sup>th</sup> October 2017



# Overview

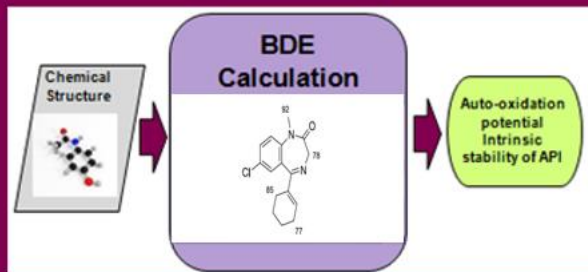
- Predictive stability tools
  - BDE tool
  - Zeneth
  - ASAP
  - Packaging predictions
- Phase I development case studies
  - IV solution
  - Tablet
  - Capsule
- Conclusions



# Predictive Stability Tools

## Bond Dissociation Energy (BDE) Tool

Predicts risk of autoxidation



## Zeneth

Predicts potential degradants and excipient interactions



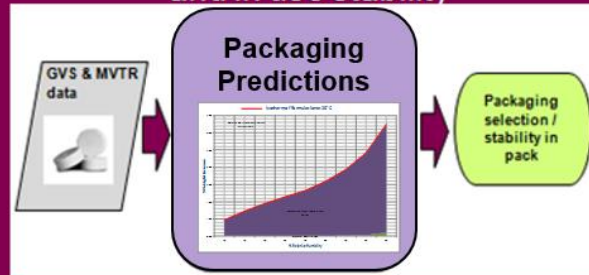
## ASAP Studies

Predict shelf life from stressed chemical stability studies



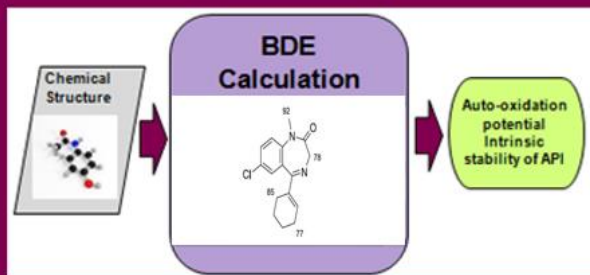
## Packaging Predictions

Predicts product water content on storage and in use stability



# Predictive Stability Tools

## Bond Dissociation Energy (BDE) Tool Predicts risk of autoxidation



- Determine risk of autoxidation
- Performed early in development
- Used to compare lead compounds
- If autoxidation risk identified, followed up with EPR spectroscopy analysis to monitor for radical formation

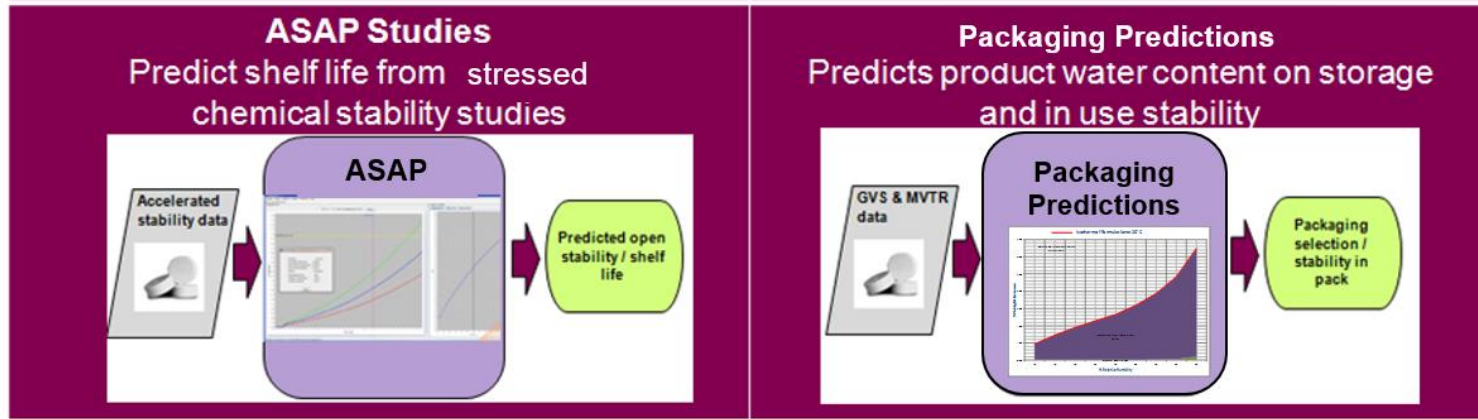
## Zeneth Predicts potential degradants and excipient interactions



- Determine potential degradants
- Performed early in development before forced degradation studies and method development
- Used to identify possible excipient incompatibilities to support formulation development



# Predictive Stability Tools



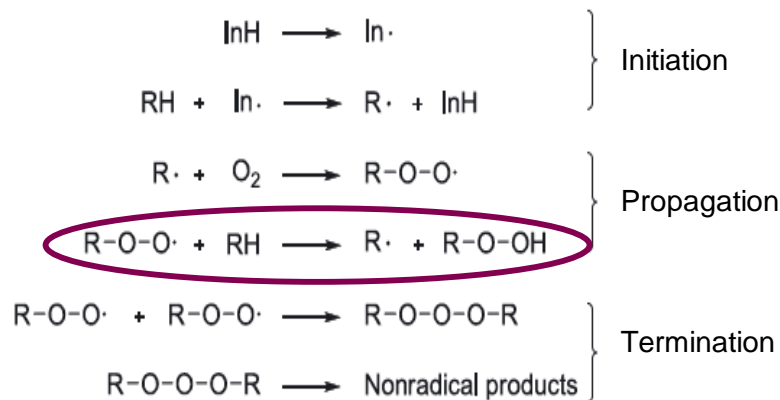
- Used throughout development to predict drug substance or product stability
- Used to support pack selection
- Determine the need for desiccant
- Set water content specifications
- Used to support pack changes during development
- Used to support pack changes post approval



# Bond Dissociation Energy Tool

The BDE values can be accurately calculated for hydrogen atoms in a molecule using relatively fast DFT methods<sup>1</sup>. The risk of autoxidation can then be estimated.

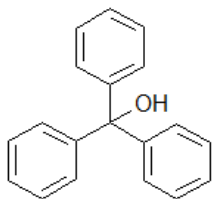
BDE < 87 kcal/mol = risk for autoxidation



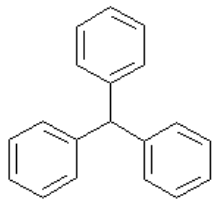
In principle, the chain reaction can be fast as long as the breaking CH-bond is weaker than the OH bond (87 kcal/mol) that is formed in the hydrogen transfer.



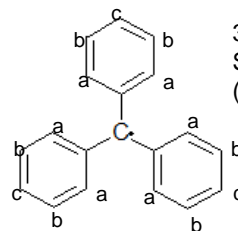
# Bond Dissociation Energy Tool



Triphenylmethanol  
BDE = 96 kcal/mol



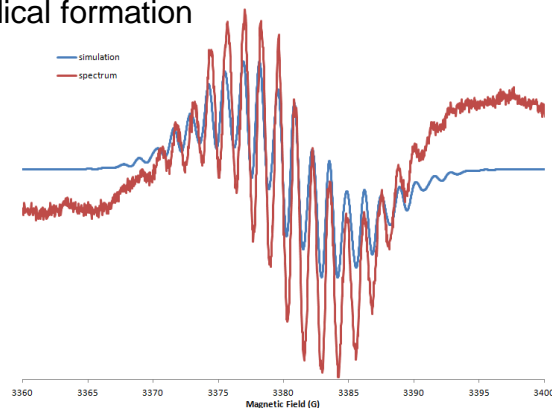
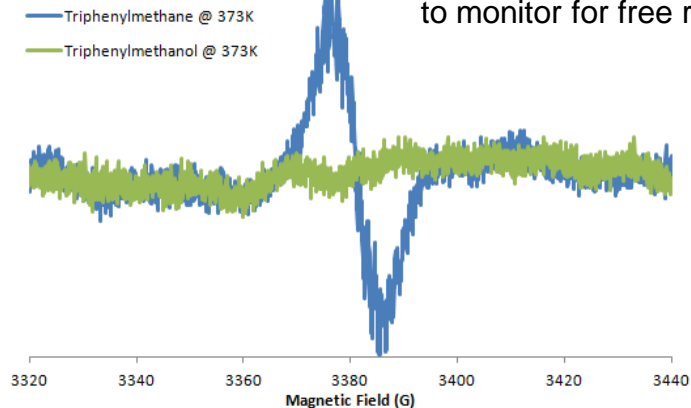
Triphenylmethane  
BDE = 72 kcal/mol



Triphenylmethane radical

3 types of non-equivalent protons  
Septet of septet of quartets  
(196 lines!)

Electron Paramagnetic Resonance spectroscopy performed  
to monitor for free radical formation



# Zeneth

Zeneth is designed to predict potential degradants of a compound from the chemical structure.

- 446 known and validated transformations (version 7.0, 2016 knowledge base)
- Data from published sources and data donated by a consortium of member organisations
- Drug Substance and Drug Product stability

Benefits of Zeneth<sup>3</sup>:

- Provides degradation information when no experimental data is available
- Data generated can be used to support regulatory submissions
- Assist in the selection of excipients, highlighting potentially problematic ones
- Helps in assignment of structure to forced degradation work
- Prevents the overlooking of possible pathways through unbiased application of transformations
- Presence or absence of conditions (light, water, oxygen, radical initiators, peroxides, metal, temperature, pH)
- Relative likelihood of competing reactions

<sup>3</sup> <http://www.lhasalimited.org/products/zeneth.htm>



# Zeneth

Example query compound

Zeneth predictions displayed in a tree layout. Zoom out for high level view & zoom in for compound specific detail

Transformation description and references to supporting literature

Simplified tree view of results

The screenshot displays the Zeneth software interface. At the top left, the 'Query Structure' window shows a chemical structure of a query compound. Below it, the 'LHASA PREDICTIONS Summary' window shows a simplified tree view with a root node 'Q' and four child nodes 'D1', 'D2', 'D3', and 'D4'. Node 'D3' is further labeled with 'I1t' and 'I2t'. The main window, 'LHASA PREDICTIONS Detail', shows a detailed tree layout. The root node is 'LIKELY', which branches into 'Hydrolysis of ester' (D1) and 'Photo-Fries rearrangement' (I1). Under 'Hydrolysis of ester', there are two sub-nodes: D1 (Formula: C<sub>9</sub>H<sub>10</sub>O<sub>2</sub>, Average Mass: 150.152, Exact Mass: 150.07113) and D2 (Formula: C<sub>9</sub>H<sub>8</sub>O<sub>2</sub>, Average Mass: 138.122, Exact Mass: 138.03169). Under 'Photo-Fries rearrangement', there are two sub-nodes: D3 (Formula: C<sub>9</sub>H<sub>10</sub>O<sub>2</sub>, Average Mass: 150.152, Exact Mass: 150.04226) and D4 (Formula: C<sub>9</sub>H<sub>8</sub>O<sub>2</sub>, Average Mass: 138.122, Exact Mass: 138.04226). A 'Transformation Description' window is open, showing the reaction:  $R_1-C(=O)-R_2 + H_2O \rightarrow R_1-C(=O)-OH + HO-R_2$ . It includes a legend: R1 = carbon or hydrogen; R2 = aliphatic carbon (not multiply bonded and not attached to another heteroatom or aromatic carbon); The carbonyl-oxygen bond is not in a ring. At the bottom, the 'LHASA PREDICTIONS Results Table' is visible.

| Parent | Parent Z | Intermediate | Degradant | Duplicate | Pathway Like | Transformation N: | Transformation Name       | Formula | Formula Gain | Formula Loss | Average M | Exact Mass | Mass Differ |
|--------|----------|--------------|-----------|-----------|--------------|-------------------|---------------------------|---------|--------------|--------------|-----------|------------|-------------|
| Q      |          | D1           |           |           | LIKELY       | 0                 |                           | C9H10O2 |              |              | 150.152   | 150.04226  |             |
| Q      |          | D2           |           |           | LIKELY       | 13                | Hydrolysis of ester       | C9H8O2  |              | C9H10O2      | 138.122   | 138.02113  | -0.20117    |
| Q      | I1t      | D3           |           |           | LIKELY       | 191               | Photo-Fries rearrangement | C9H10O2 |              |              | 150.152   | 150.04226  | 0.00000     |
| Q      | I2t      | D4           |           |           | LIKELY       | 191               | Photo-Fries rearrangement | C9H8O2  |              |              | 138.122   | 138.04226  | 0.00000     |

Zeneth also includes a list of common excipients and is able to predict whether they are likely to interact with the query compound

Table to show all degradants generated. Can be filtered upon molecular formula or molecular mass



# ASAP Studies

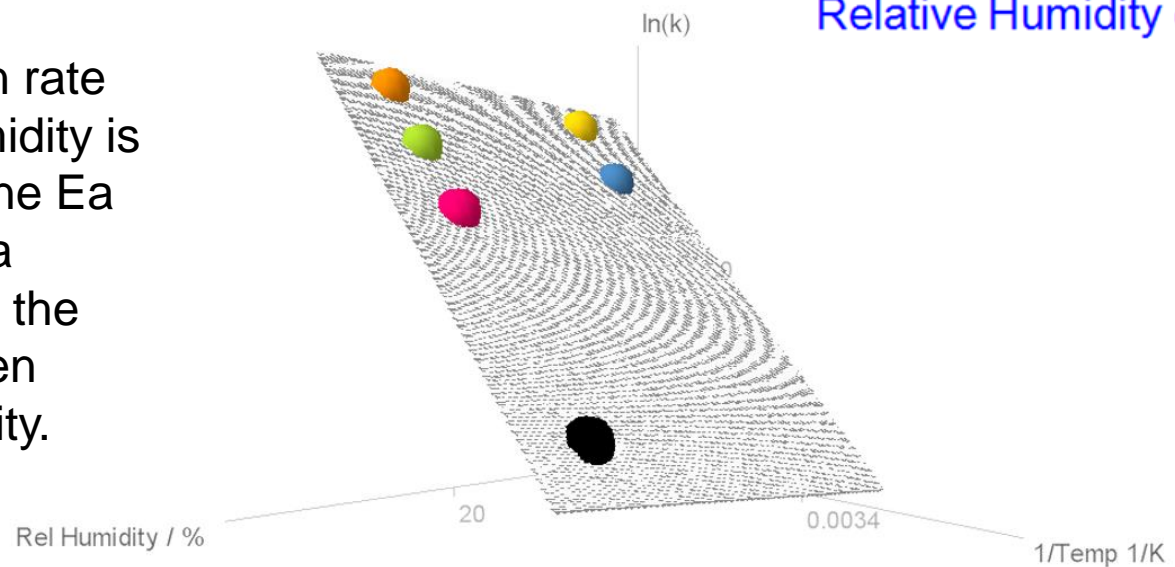
Based on the Arrhenius equation modified for solid state degradation

Sensitivity to moisture

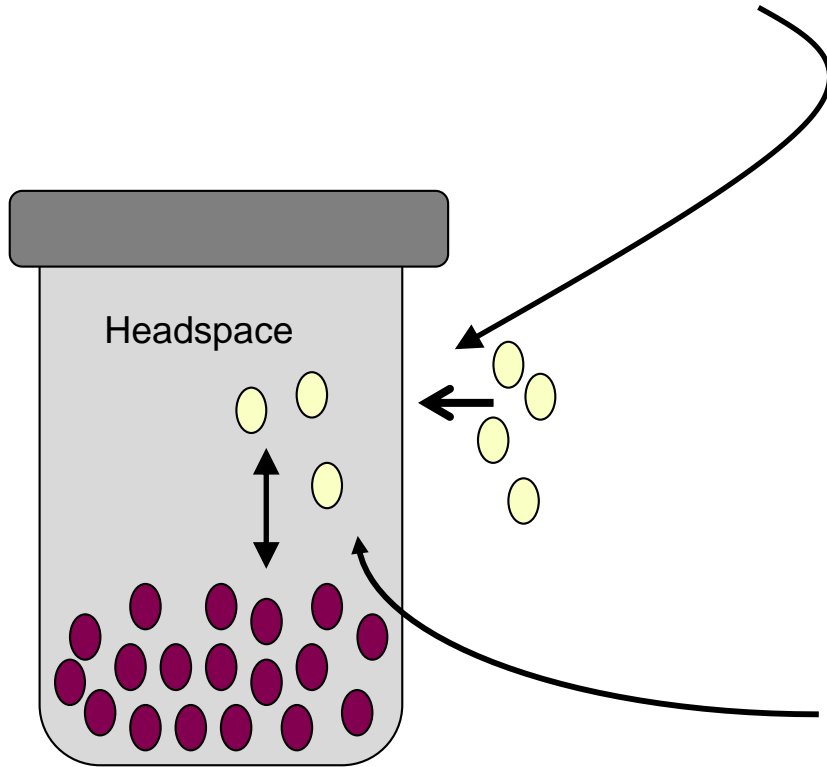
$$\ln(k) = \ln(A) - E_a/RT + (B \times RH)$$

Relative Humidity (%)

If the change in reaction rate with temperature & humidity is measured, can determine  $E_a$  and  $\ln(A)$  and  $B$  and via extrapolation determine the reaction rate at any given temperature and humidity.

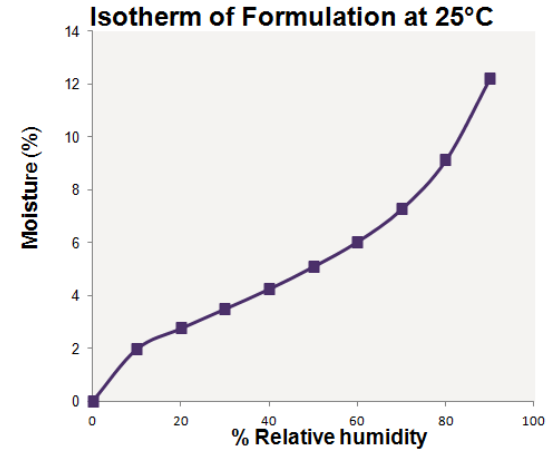


# Packaging Predictions



Depends on permeation of the barrier

Difference in partial pressure of water between ambient environment and headspace



Equilibrium between headspace and tablets is defined by the Gravimetric Vapour Sorption isotherm of the tablets



# Phase I formulation development - IV solution

## Applications of predictive stability tools

1. Drug substance – BDE and Zeneth predictions
2. Drug substance - ASAP study
3. Drug product – Zeneth predictions, excipient incompatibilities
4. Drug product prototype screening – ASAP
5. Drug product shelf life prediction – ASAP
6. Regulatory Applications



# Phase I formulation development - IV solution

## 1. Drug substance – BDE and Zeneth predictions

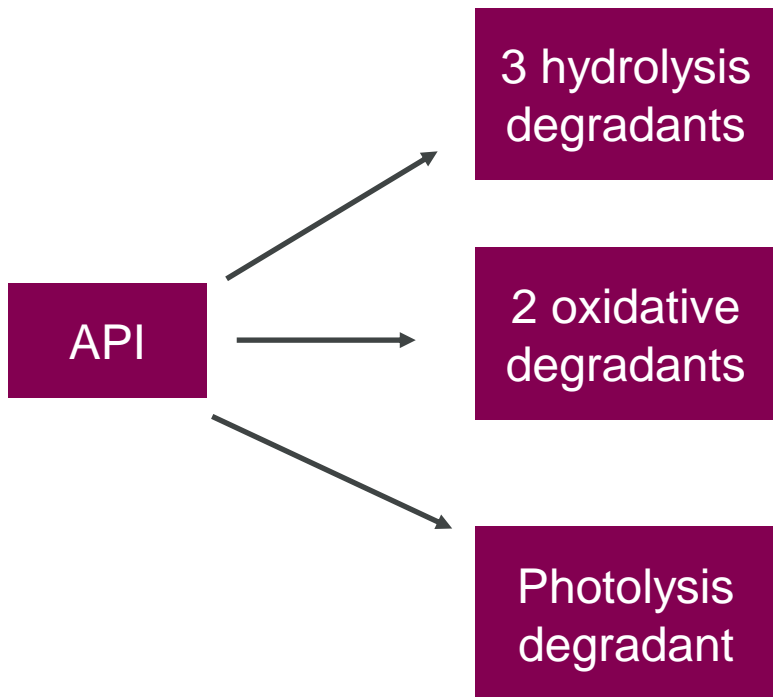
### Zeneth Conditions

- 20°C, pH 1, 4, 7, 10 and 14 in presence of water, peroxides, radical initiators, metals, oxygen and light
- 50°C, pH 1, 7 and 14 in presence of water, peroxides, radical initiators, metals, oxygen and light
- A second screen in the presence of formaldehyde was performed
- Only degradants deemed likely or highly likely were included
- To simplify the screen secondary degradants were only investigated when deemed very likely
- Similarly degradants were not allowed to react with themselves



# Phase I formulation development - IV solution

## 1. Drug substance – BDE and Zeneth predictions



Lowest BDE = 84 kcal/mol

In line with one of Zeneth oxidation predictions



# Phase I formulation development - IV solution

## 2. Drug substance ASAP study

| Temperature (°C) | Humidity (% RH) | Storage Time (weeks) |
|------------------|-----------------|----------------------|
| Initial          | Initial         | 0 (3 repeats), X     |
| 50               | 75              | 3, Sx2               |
| 60               | 11              | 3, Sx2               |
| 60               | 75              | 3 (5 repeats), Sx2   |
| 70               | 11              | 3, Sx2, C, X         |
| 70               | 75              | 3, Sx2, C, X         |
| 80               | 30              | 3, S, C, X           |

Impurities analysis by LC at all time points  
S = spare samples, pulled after 8 and 12 weeks  
X = XRPD sample, C = sample for chiral testing

No degradation was observed



# Phase I formulation development - IV solution

## 3. Drug product – Zeneth predictions, excipient incompatibilities

No interactions were predicted by Zeneth between the API and any of the excipients or known excipient impurities.

Zeneth did not predict a peroxide mediated degradation mechanism for the API.

Based on these predictions there was a low risk of an API:excipient interaction in this formulation.

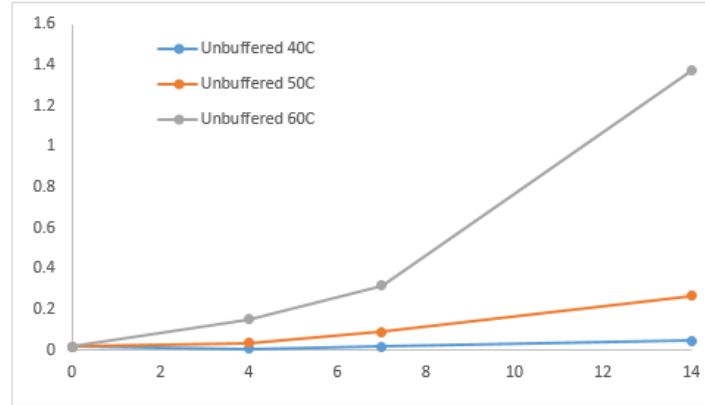
However API degradation is known to be pH dependant.



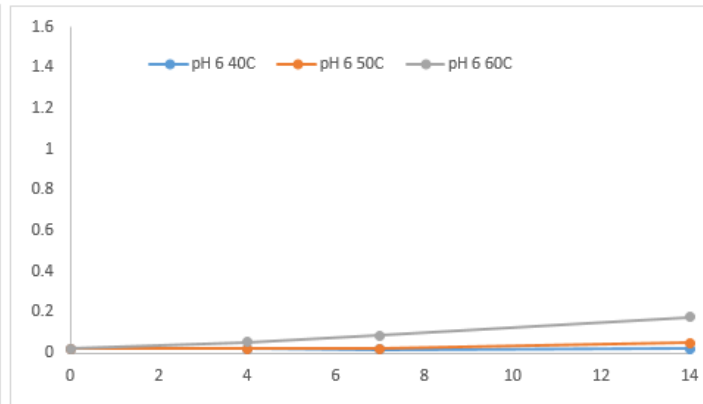
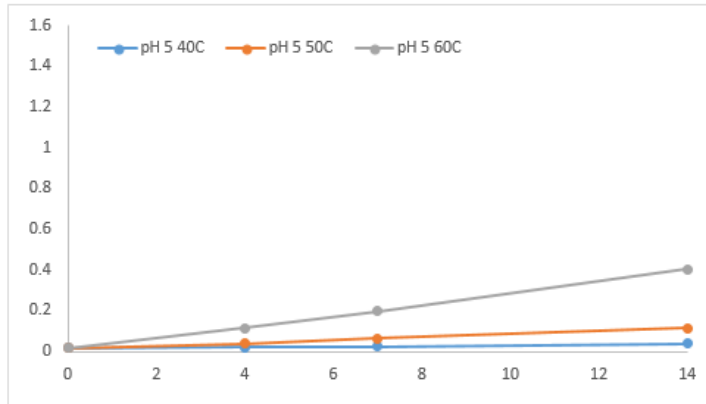
# Phase I formulation development - IV solution

## 4. Drug product prototype screening – ASAP

Three formulations screened; unbuffered, pH 5 and pH 6  
One main hydrolysis degradant observed (previously predicted by Zeneth)

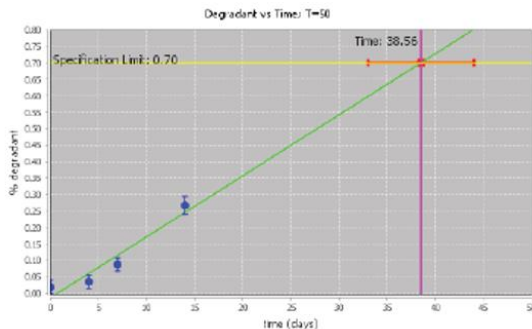
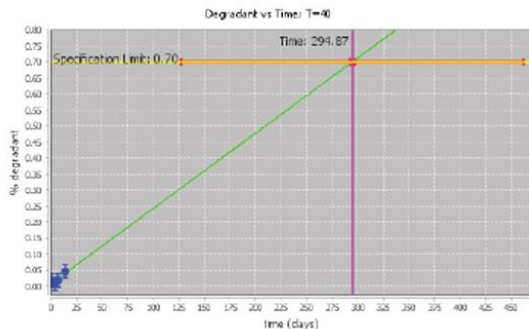


Unbuffered solution gave most degradation followed by pH 5 and then pH 6 the least degradation

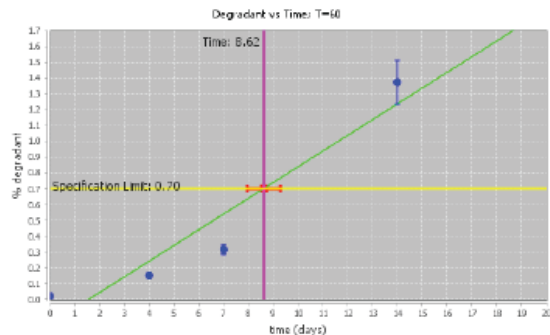


# Phase I formulation development - IV solution

## 4. Drug product prototype screening – ASAP



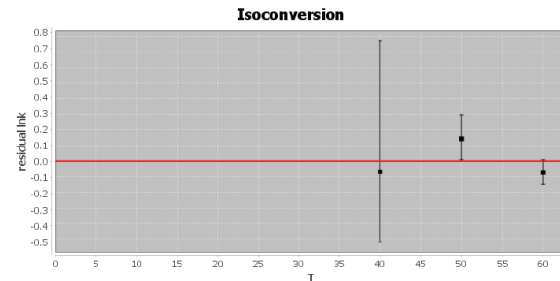
**Unbuffered solution results**  
Assuming linear fit to the degradation kinetics



Arrhenius Data

|                |        |              |
|----------------|--------|--------------|
| lnA            | 52.9 ± | 8.8          |
| E <sub>a</sub> | 36.6 ± | 5.8 kcal/mol |
| B              | -      | ±            |
| R <sup>2</sup> | 1.00   |              |
| Q <sup>2</sup> | 0.93   |              |

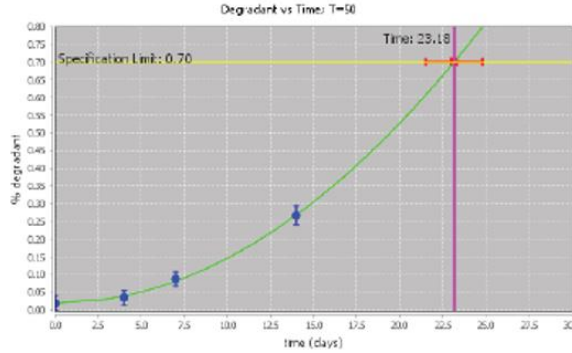
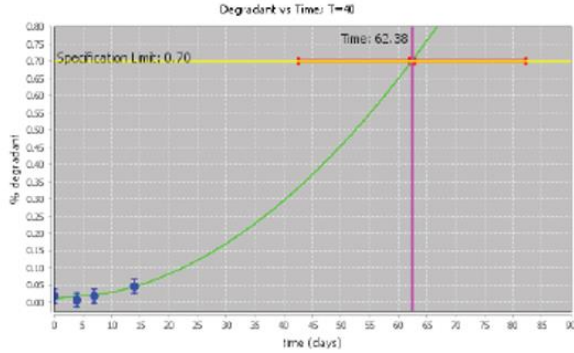
Residual Plot



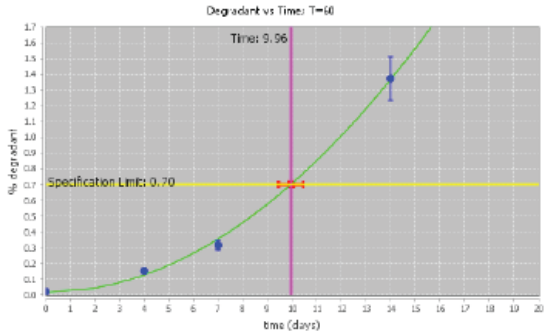
A “good” model was generated, although limited degradation at some conditions.

# Phase I formulation development - IV solution

## 4. Drug product prototype screening – ASAP

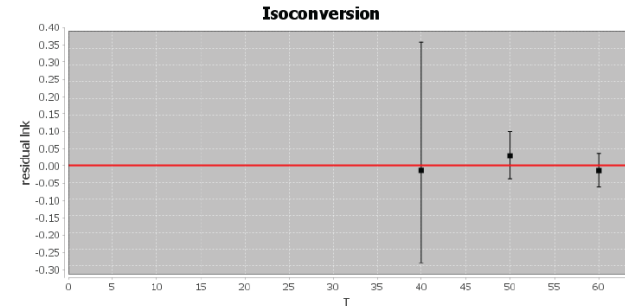


**Unbuffered solution results**  
Assuming non-linear fit to the degradation kinetics



|                |        |              |
|----------------|--------|--------------|
| InA            | 26.1 ± | 5.9          |
| E <sub>a</sub> | 19.0 ± | 3.9 kcal/mol |
| B              | -      | -            |
| R <sup>2</sup> | 1.00   |              |
| Q <sup>2</sup> | 0.99   |              |

Residual Plot



A “good” model was generated, although limited degradation at some conditions.

# Phase I formulation development - IV solution

## 4. Drug product prototype screening – ASAP

Predictions at 25°C/60% RH

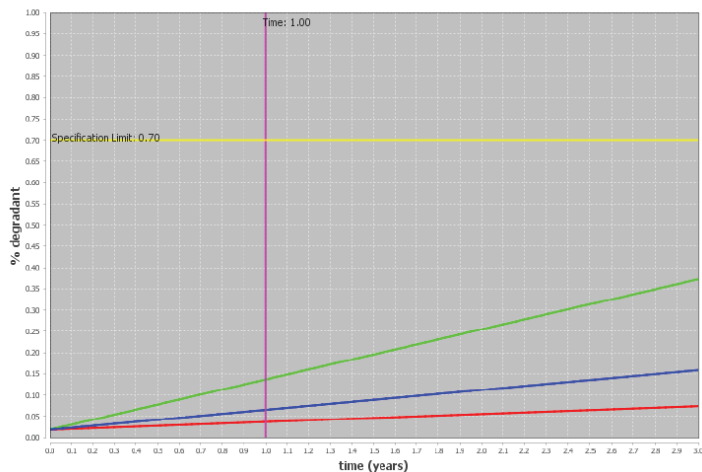
Linear fit

|                        |        |       |
|------------------------|--------|-------|
| Spec Limit             | 0.70   |       |
| Median Shelf-Life      | > 3.00 | years |
| Probability of Passing | 99.81  | %     |

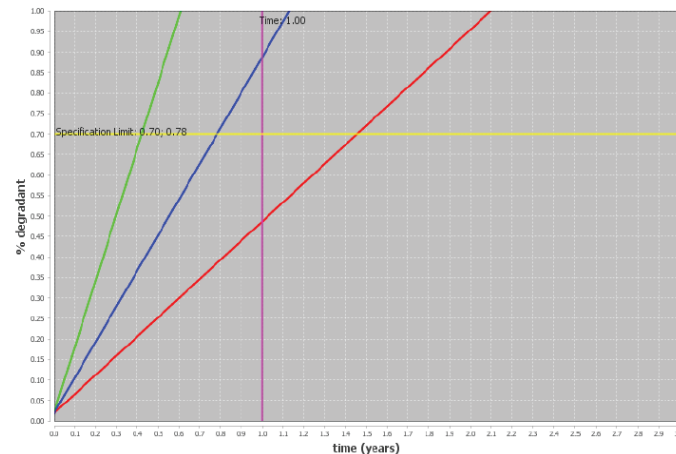
Non-linear fit

|                        |       |       |
|------------------------|-------|-------|
| Spec Limit             | 0.70  |       |
| Median Shelf-Life      | 0.78  | years |
| Probability of Passing | 34.80 | %     |

Degradant/Time Plot

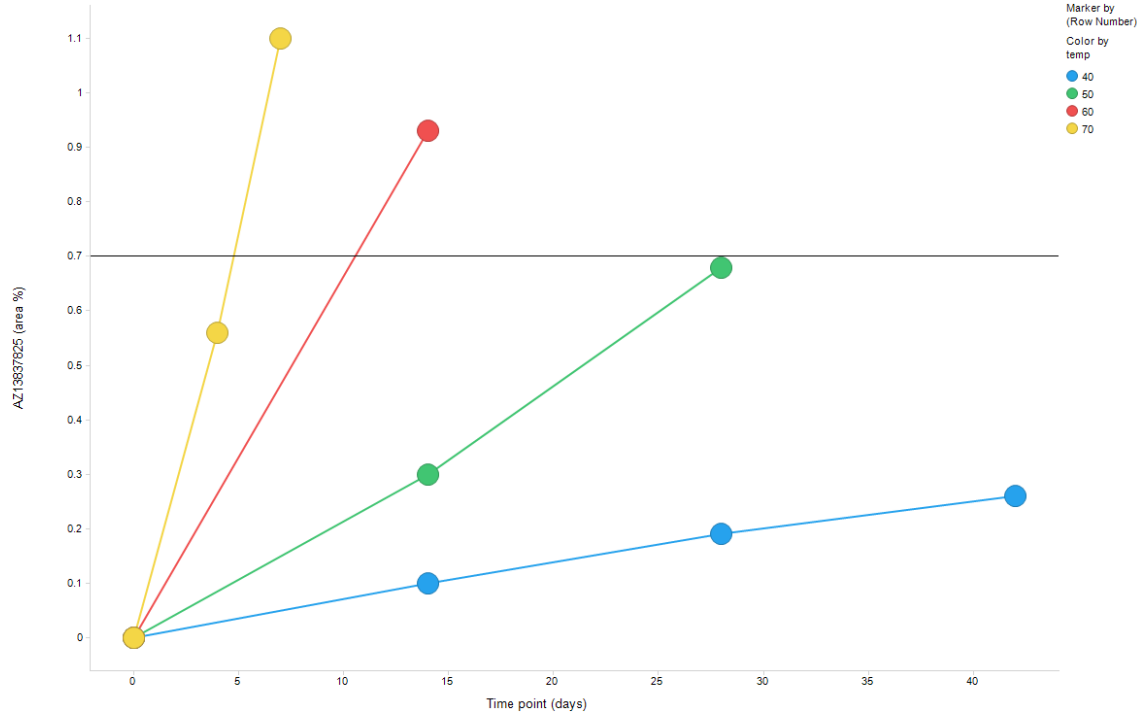


Degradant/Time Plot



# Phase I formulation development - IV solution

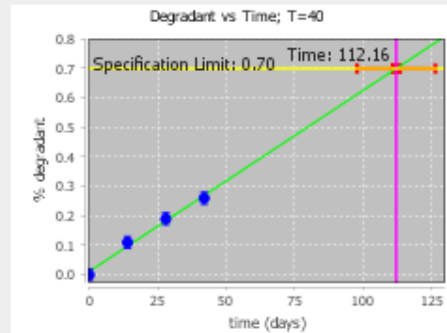
## 5. Drug product (pH 5 buffer) shelf life prediction – ASAP



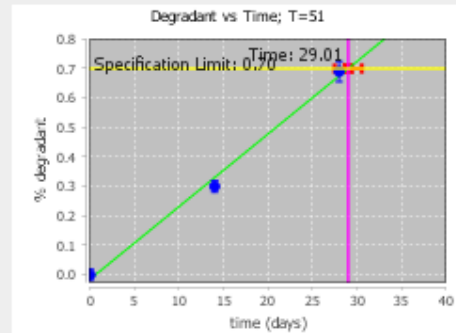
# Phase I formulation development - IV solution

## 5. Drug product (pH 5 buffer) shelf life prediction – ASAP

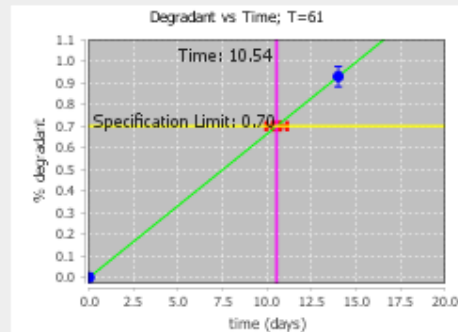
Isoconversion Ratio: 0.37  
Extrapolation Used



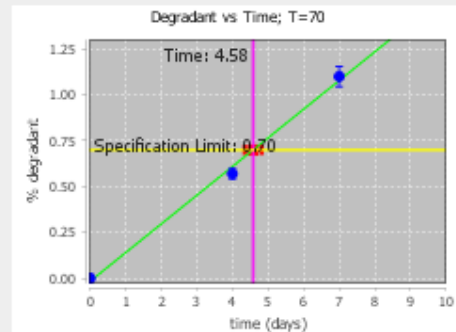
Isoconversion Ratio: 0.97



Isoconversion Ratio: 1.33



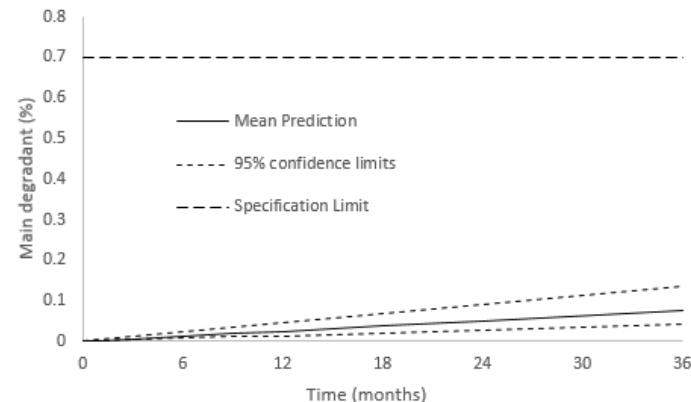
Isoconversion Ratio: 1.53



Arrhenius Data

|                |        |              |
|----------------|--------|--------------|
| lnA            | 31.3 ± | 1.4          |
| E <sub>a</sub> | 22.6 ± | 0.9 kcal/mol |
| B              | -      | -            |
| R <sup>2</sup> | 1.00   |              |
| Q <sup>2</sup> | 0.98   |              |

“Good” model  
3 year shelf life prediction at 5°C



# Phase I formulation development - IV solution

## 6. Regulatory Applications

- Drug substance and drug product ASAP data was presented in the Phase I regulatory submission to support a 12 month shelf life/retest period, in the absence of long term stability data
- The MHRA requested ICH compliant stability data for both the substance and product
- There were no questions from Germany on the drug substance stability section but they queried the shelf life claim for the drug product
- The submission was accepted with no questions in Holland



# Phase I formulation development - tablet

## Applications of predictive stability tools

1. Drug substance – BDE and Zeneth predictions –  
4 main degradants predicted
2. Drug substance ASAP study – minimal degradation detected
3. Drug product – Zeneth predictions, excipient incompatibilities -  
no significant risk identified
4. Drug product shelf life prediction – ASAP
5. Regulatory Applications



# Phase I formulation development - tablet

## 4. Drug product shelf life prediction – ASAP

ASAP studies for 2 formulation strengths, 5 and 50 mg

Protocol for impurities and appearance

| Temperature (°C) | Humidity (%RH) | Storage Time (weeks) |
|------------------|----------------|----------------------|
| Initial          | Initial        | 0                    |
| 50               | 75             | 3, 6, S              |
| 60               | 30             | 3, 6, S              |
| 60               | 75             | 3, 6, S              |
| 70               | 11             | 2, 3, 6, S           |
| 70               | 75             | 2, 3, 6, S           |
| 80               | 30             | 2, 3, 6, S           |

S = spare sample

Protocol for dissolution

| Temperature (°C) | Humidity (%RH) | Storage Time (weeks) |
|------------------|----------------|----------------------|
| Initial          | Initial        | 0                    |
| 40               | 11             | [4]                  |
| 40               | 55             | [4]                  |
| 40               | 75             | 4                    |

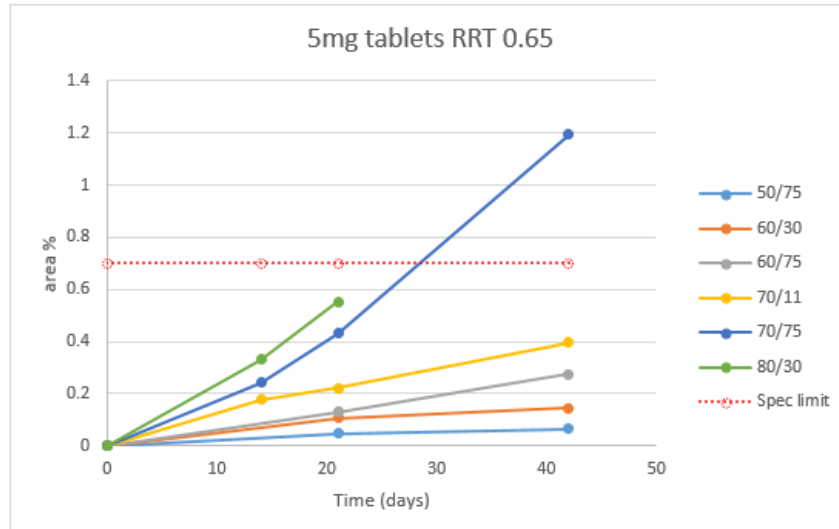
[ ] = optional



# Phase I formulation development - tablet

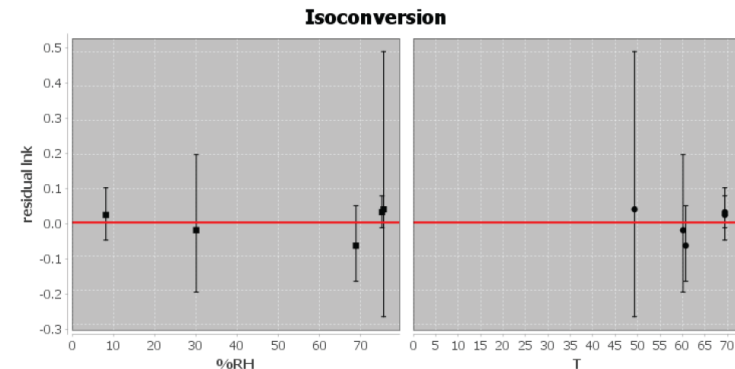
## 4. Drug product shelf life prediction – ASAP

No change in dissolution profile or appearance during study



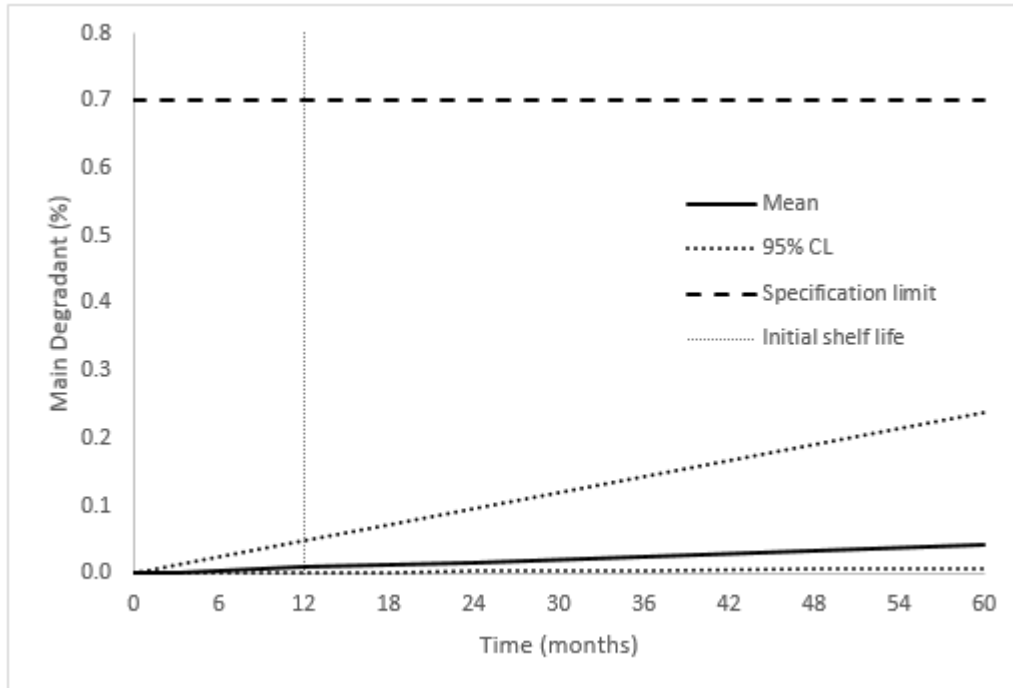
|                |          |              |
|----------------|----------|--------------|
| lnA            | 40.5 ±   | 6.2          |
| E <sub>a</sub> | 30.8 ±   | 4.2 kcal/mol |
| B              | 0.0147 ± | 0.0021       |
| R <sup>2</sup> | 1.00     |              |
| Q <sup>2</sup> | 0.98     |              |

“Good” model



# Phase I formulation development - tablet

## 4. Drug product shelf life prediction – ASAP

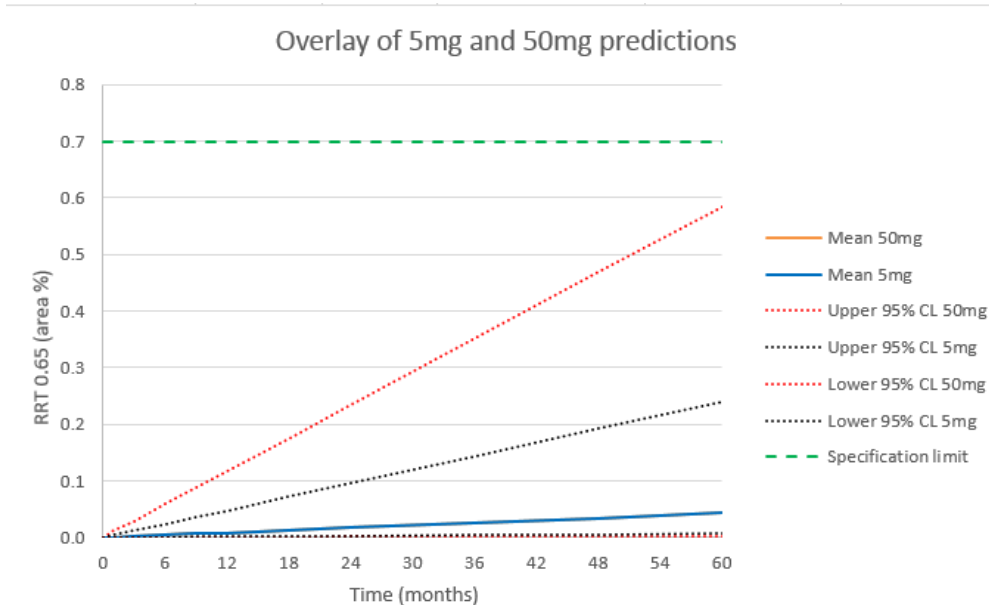


Predictions at 25°C/60% RH support an initial shelf life of 12 months but suggest the product is likely to achieve 5 years



# Phase I formulation development - tablet

## 4. Drug product shelf life prediction – ASAP



- When comparing both strengths of tablets, it can be seen that the error for the 50mg tablets is much larger than for the 5mg
- This is because the 50mg tablets did not degrade as much as the 5mg, therefore increased extrapolation required to reach specification



# Phase I formulation development - tablet

## 5. Regulatory Applications

- Drug substance and drug product ASAP data was presented in the Phase I regulatory submission to support a 12 month shelf life/retest period, in the absence of long term stability data
- Submitted to UK and USA
- Waiting for questions



# Phase I formulation re-development - capsule

## Applications of predictive stability tools

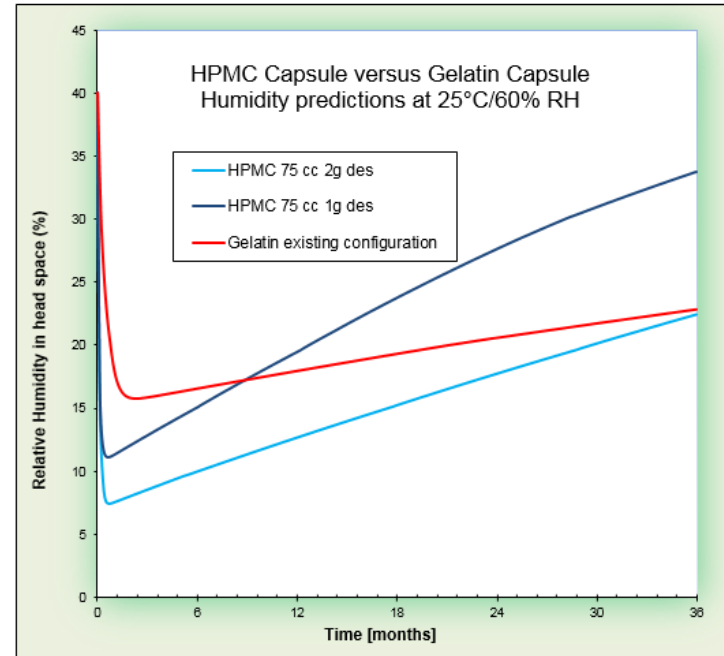
1. Packaging predictions drug product
2. Drug product shelf life prediction – ASAP
3. Regulatory Applications



# Phase I formulation re-development - capsule

## 1. Packaging predictions

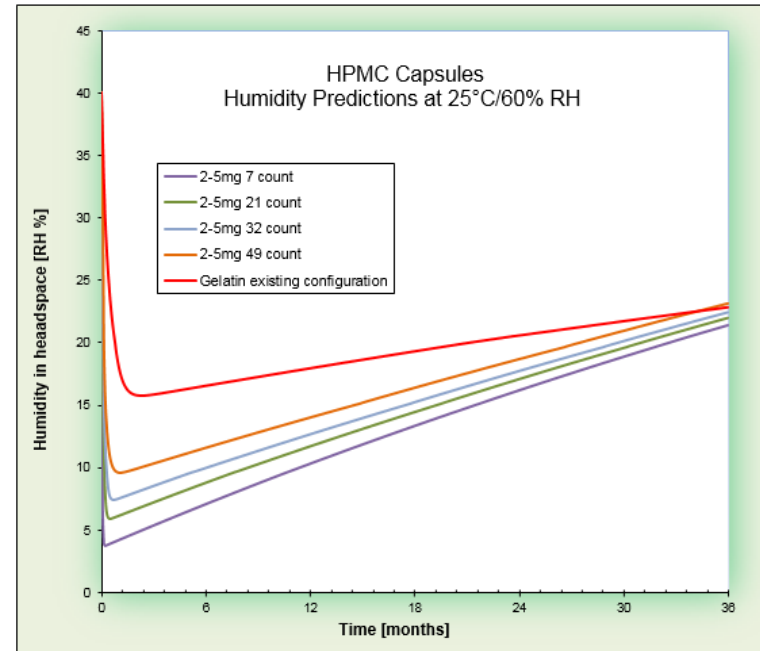
- Reformulation from Gelatin capsules to HPMC capsules
- Performed packaging predictions to determine humidity in bottle for existing gelatin formulation and to predict humidity for the HPMC capsules in a new bottle and decide on the level of desiccant required



# Phase I formulation re-development - capsule

## 1. Packaging predictions

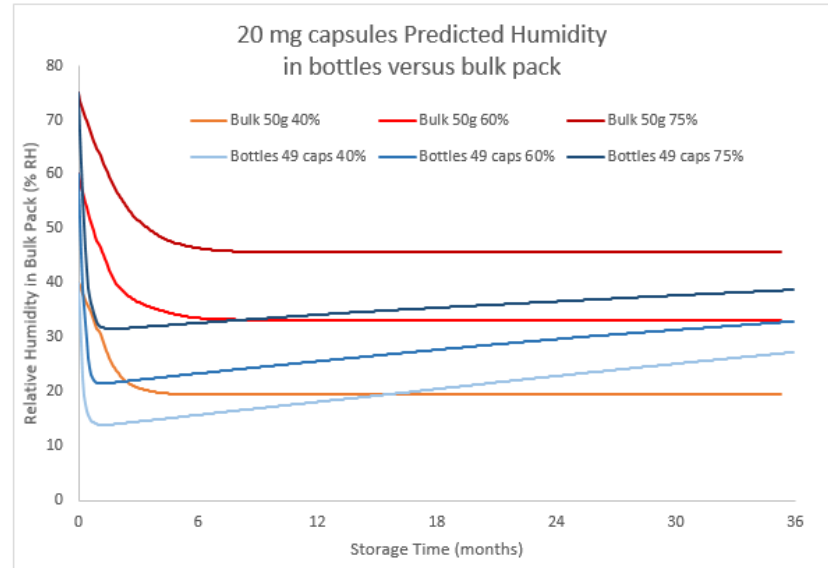
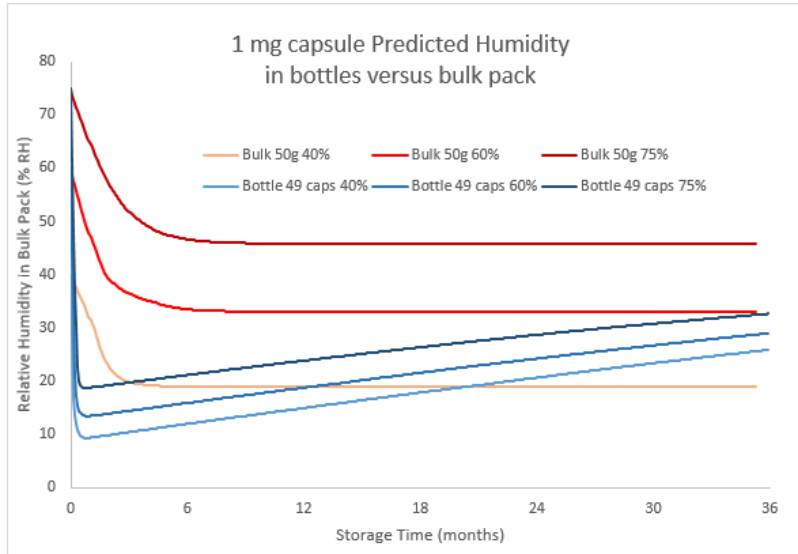
- Reformulation from Gelatin capsules to HPMC capsules
- Performed packaging predictions to determine humidity in bottle for existing gelatin formulation and to predict humidity for the HPMC capsules in a new bottle and decide on the level of desiccant required
- Also predicted the impact of changing the capsule count



# Phase I formulation re-development - capsule

## 1. Packaging predictions

Packaging predictions to compare the bottle and bulk packs were also performed, bottles containing 49 tablets and the bulk pack with 50 g desiccant.



When packing at 60 or 75% RH the bulk pack is less protective than the bottle.  
33 At 40% the packs are more equivalent over the 3 year period.



# Phase I formulation re-development - capsule

## 2. Drug product shelf life prediction – ASAP

- Two formulation strengths, 1 mg and 20 mg

| Temperature (°C) | Humidity (% RH) | Storage Time (weeks) |
|------------------|-----------------|----------------------|
| Initial          | Initial         | 0 (3 repeats), X     |
| 50               | 30              | 2, 4, S              |
| 50               | 75              | 2, 4, S              |
| 60               | 30              | 2, 4 (5 repeats), S  |
| 60               | 75              | 2, 4, S              |
| 70               | 11              | 1, 2, S, C, X        |
| 70               | 75              | 1, 2, S, C, X        |

Impurities analysis by LC at all time points (2 capsules per analysis)

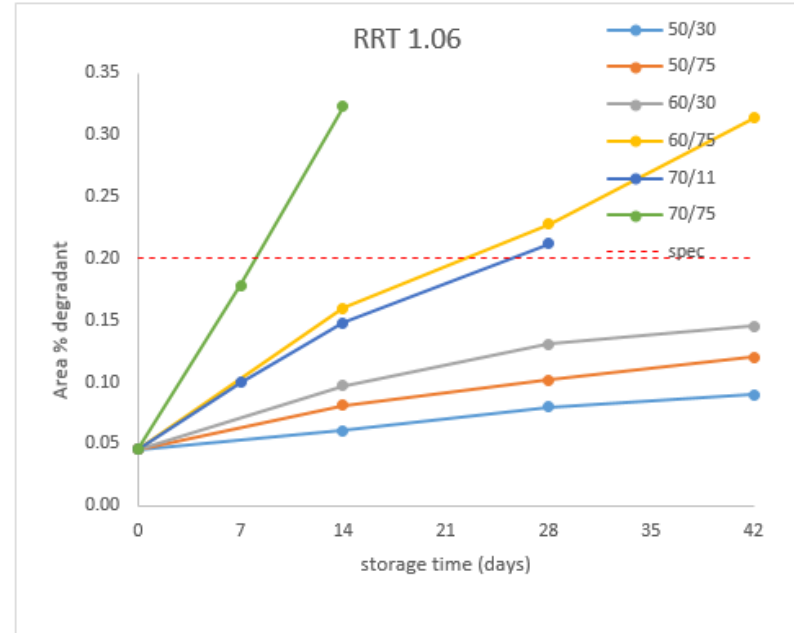
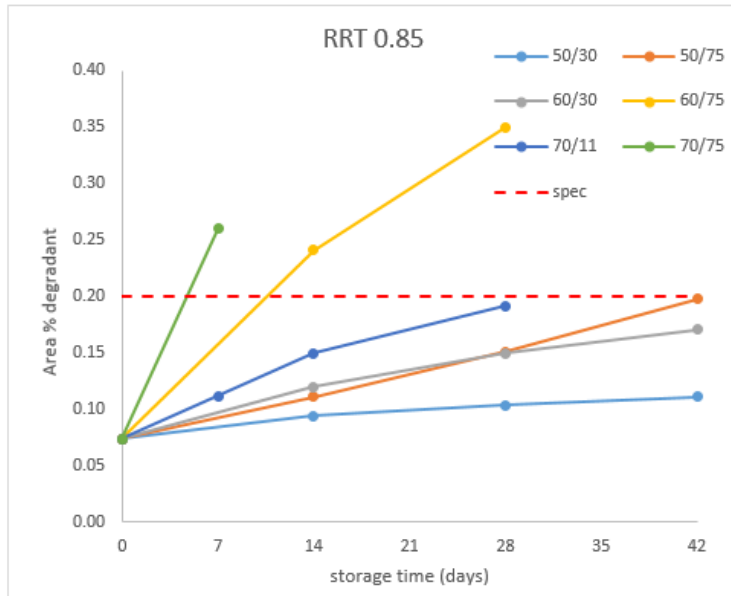
S = spare sample, X = XRPD sample, C = chiral sample



# Phase I formulation re-development - capsule

## 2. Drug product shelf life prediction – ASAP

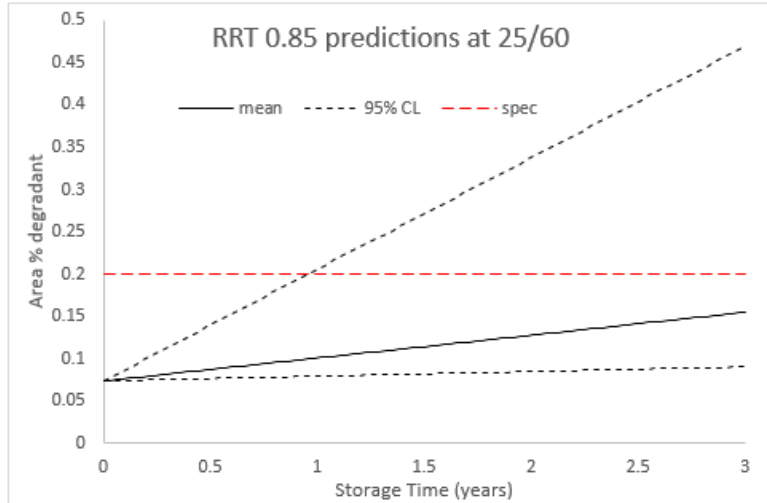
Main degradants identified as RRT 0.85 and RRT 1.06, both with an initial specification limit of < 0.2%



# Phase I formulation re-development - capsule

## 2. Drug product shelf life prediction – ASAP

RRT 0.85 – 3 year shelf life based on mean prediction. The upper 95% confidence limit of the predictions just falls short of 1 year shelf life at 25/60 open.



A “good” model was generated for RRT 0.85, with  $R^2$  and  $Q^2$  terms  $> 0.9$  and no evidence of outliers (all residuals crossing the line).

In total, ran 8 ASAP models across two formulation strengths, but RRT 0.85 was determine to be the shelf life limiting attribute.

36 Based on the ASAP study the specification limit for this impurity was increased to 0.7%. Higher humidity potentially in bulk pack no longer an issue.



# Phase I formulation re-development - capsule

## 3. Regulatory Applications

- HPMC capsule ASAP data was presented in the IND, alongside 1 months long term stability data
- Submitted to USA
- Accepted without questions
- ASAP data was used to support an internal 12 month shelf life for the bottle pack configuration and a 6 month shelf life in the bulk pack



# Conclusions

- Predictive stability tools can be used to support development
- Zeneth and BDE tools can aid understanding of potential degradation mechanisms across both substance and product
- ASAP and packaging predictions are powerful tools to predict chemical degradation, water content and pack humidity and can influence pack selection, pack configuration, storage conditions, shelf life/retest period claims, specification setting and control strategy
- Regulatory acceptance of predicted stability data is growing



# Acknowledgements

Faye Turner  
Andrew Brookes  
Carolyn Gordon  
Jonathan Bright  
Emily Roddy  
Pam Harrison  
Thomas Andersson  
Per-Ola Norrby  
Ian Ashworth  
David Benstead  
Magnus Fransson  
Johan Remmelgas  
Nadim Akhtar

Keith Parker  
John Nightingale  
Angela Jordan  
Dawn Adkin  
Angela Currie  
Darren Gore  
Andrew Phillips  
Ben McKeever-Abbas  
Andrew Poulton  
Paul Cronin

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